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Desolvation and Dehydrogenation of Solvated Magnesium Salts of Dodecahydrododecaborate: Relationship between Structure and Thermal Decomposition

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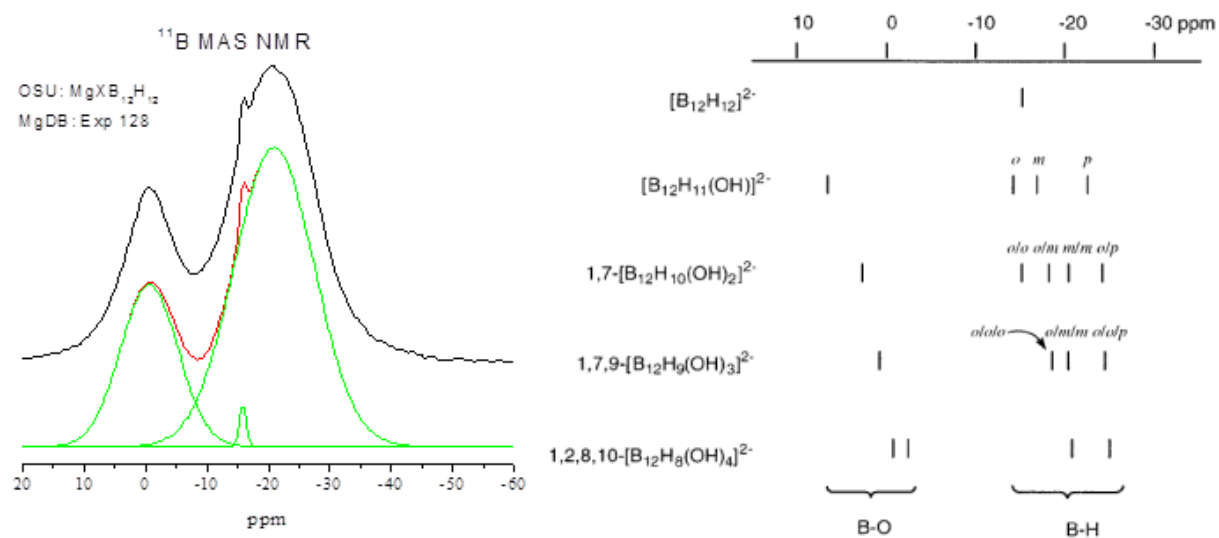


Figure S1. Solid state ^{11}B MAS NMR spectra of dehydrated residual (black) and its decomposition (green) (left); solution ^{11}B NMR spectra of different substitution $\text{B}_{12}\text{H}_{(12-x)}\text{X}_x^{2-}$ (right).^{S1}

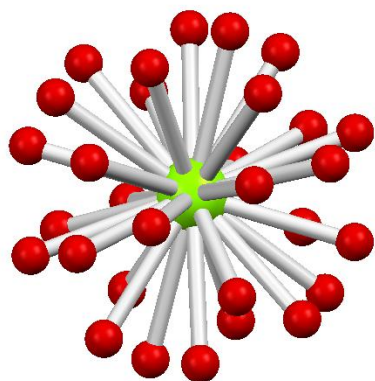


Figure S2. Magnesium coordinated by severely disordered water molecules in $\text{Mg}(\text{H}_2\text{O})_6\text{B}_{12}\text{H}_{12}\cdot 6\text{H}_2\text{O}$.

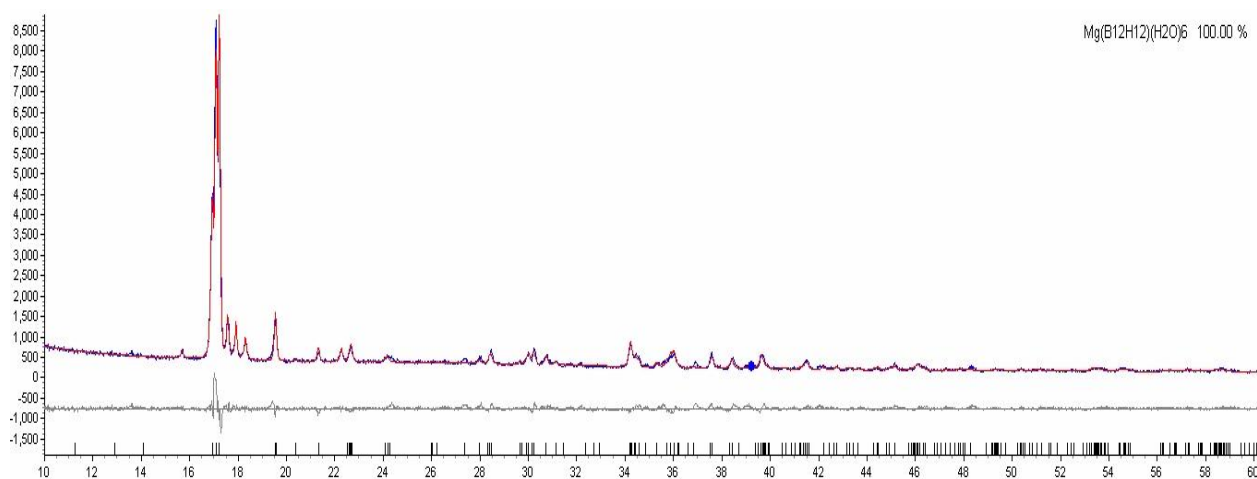
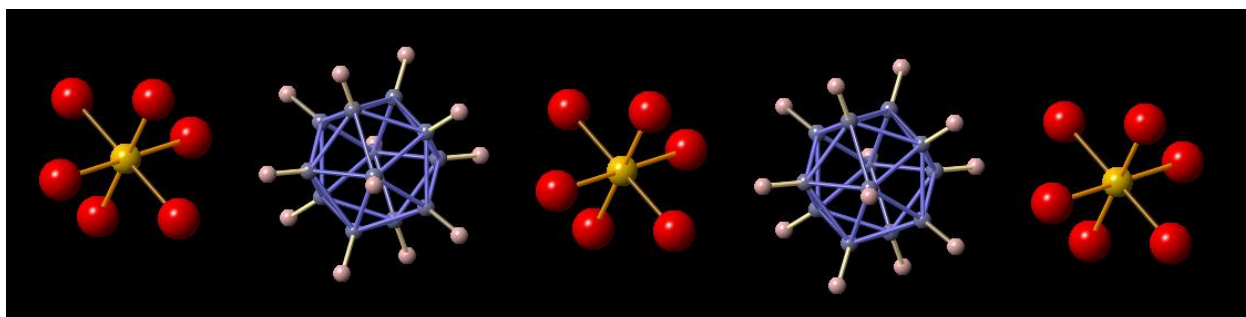


Figure S3. Powder x-ray diffraction pattern Rietveld refinement of $\text{Mg}(\text{H}_2\text{O})_6(\text{B}_{12}\text{H}_{12})$ ($\lambda=1.540596\text{\AA}$). Observed data, calculated patterns, and the difference curves are plotted as a blue line, a red line, and a grey line, respectively. The 2-Theta reflections are shown in blue.

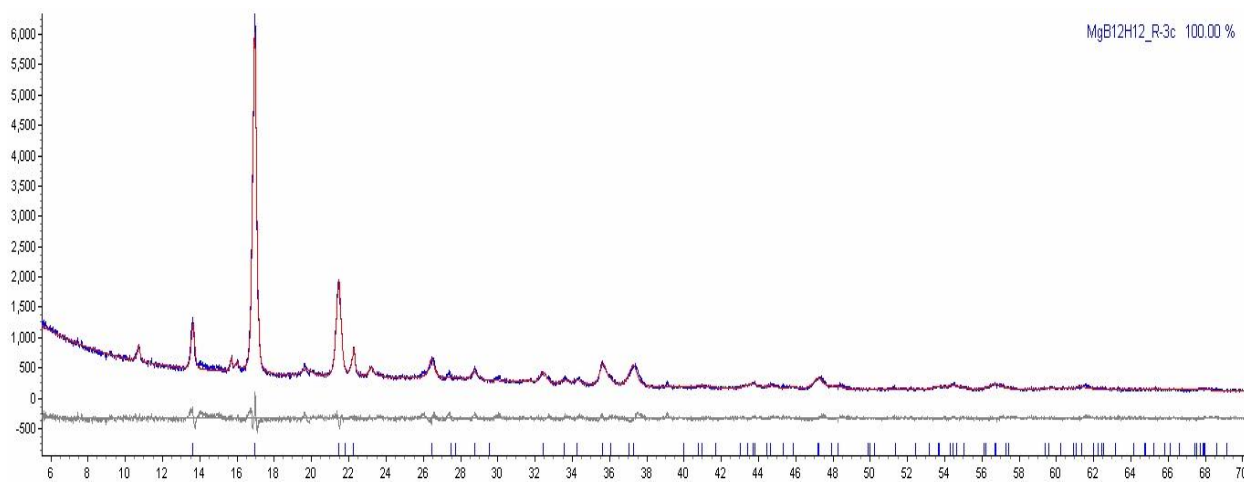
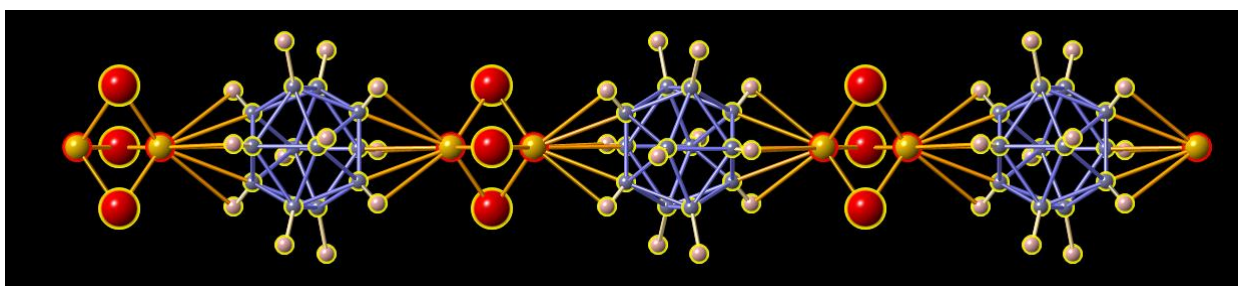


Figure S4. Powder x-ray diffraction pattern Rietveld refinement of $\text{Mg}(\text{H}_2\text{O})_3(\text{B}_{12}\text{H}_{12})$ ($\lambda=1.540596\text{\AA}$). Observed data, calculated patterns, and the difference curves are plotted as a blue line, a red line, and a grey line, respectively. The 2-Theta reflections are shown in blue.

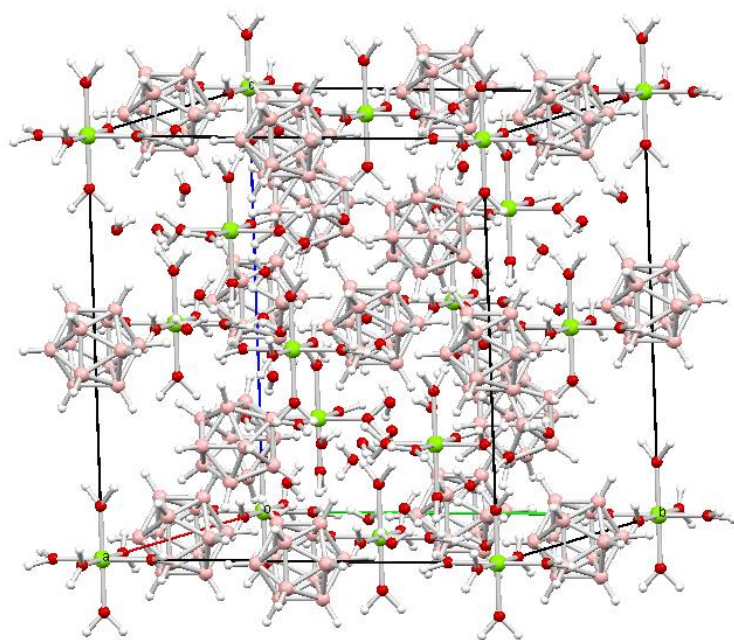


Figure S5. Crystal structure of $\text{Mg}(\text{H}_2\text{O})_6\text{B}_{12}\text{H}_{12} \cdot 6\text{H}_2\text{O}$ previously determined at room temperature.^{S2} pink, B; green, Mg; red, O. white, H.

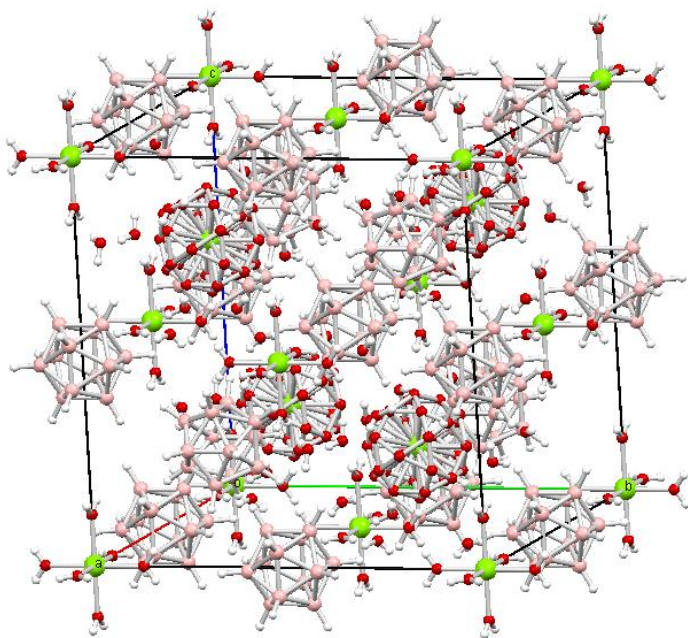


Figure S6. Crystal structure of $\text{Mg}(\text{H}_2\text{O})_6\text{B}_{12}\text{H}_{12}\cdot 6\text{H}_2\text{O}$ re-determined at 150K. pink, B; green, Mg; red, O. white, H.

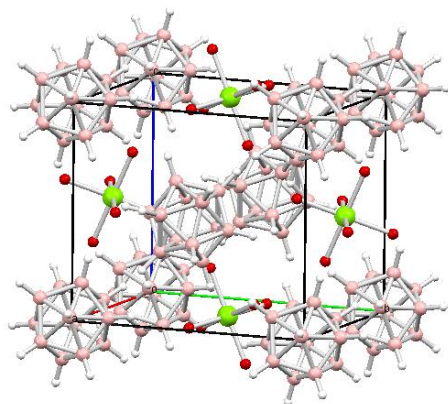


Figure S7. Crystal structure of $\text{Mg}(\text{H}_2\text{O})_6\text{B}_{12}\text{H}_{12}$. pink, B; green, Mg; red, O. white, H.

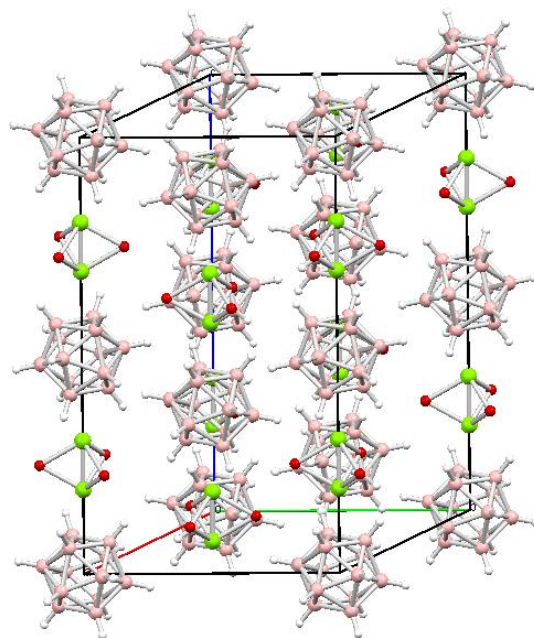


Figure S8. Crystal structure of $\text{Mg}(\text{H}_2\text{O})_3\text{B}_{12}\text{H}_{12}$. pink, B; green, Mg; red, O. white, H.

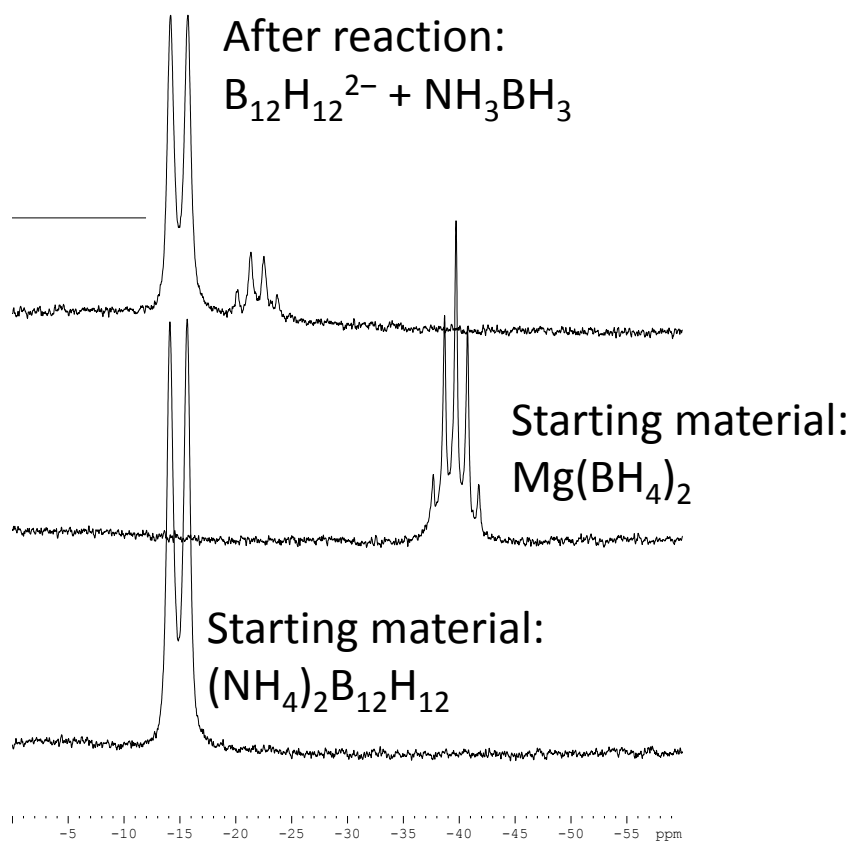


Figure S9. ^{11}B NMR spectra monitoring the reaction of $\text{Mg}(\text{BH}_4)_2 + (\text{NH}_4)_2\text{B}_{12}\text{H}_{12}$ in THF

Table S1. Crystallographic details for Mg(H₂O)₆B₁₂H₁₂·6H₂O and Mg(CH₃OH)₆B₁₂H₁₂·6CH₃OH

formula	B ₁₂ H ₃₆ MgO ₁₂	C ₁₂ H ₆₀ B ₁₂ MgO ₁₂
formula weight	382.32	550.63
crystal system	cubic	trigonal
space group	F23	R-3c
Z	8	6
a, Å	16.3149(3)	13.939(2)
c, Å		29.833(6)
T, K	150(2)K	150(2)
V, Å ³	4342.63(14)	5019.8(14)
D _{calc'd} (g·cm ⁻³)	1.170	1.093
μ (mm ⁻¹)	0.121	0.097
λ, Å	0.71073	0.71073
no. reflns collected	12944	4824
no. unique reflns	838 (Rint = 0.034)	1289 (Rint = 0.0196)
R1 [I > 2σ (I)] ^a	0.0313	0.0383
wR2(all data) ^b	0.0803	0.1200

$$^{[a]} R1 = \sum ||F_o| - |F_c|| / \sum |F_o| \quad ^{[b]} wR2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Table S2 TOPAS Rietveld Refinement results for Mg(H₂O)₆B₁₂H₁₂.

Empirical formula	B ₁₂ H ₂₄ O ₆ Mg
<i>F</i> _w	274.22
Space group	P2 ₁ /c (No. 14)
<i>a</i> (Å)	7.855(1)
<i>b</i> (Å)	10.460(1)
<i>c</i> (Å)	9.077 (1)
β (°)	90.62(1)
<i>V</i> (Å ³)	745.75(14)
<i>Z</i>	2
<i>T</i> (K)	295
λ (Å)	1.540596
ρ_{cal} (g cm ⁻³)	1.16740(21)
pattern range (2θ , °)	10 to 60
step size (2θ , °)	0.0173
step scan time (s)	0.1
no. of contributing reflns	2899
<i>R</i> _{wp}	0.0853
<i>R</i> _p	0.0637
<i>R</i> (<i>F</i> ²)	0.1248

Table S3 TOPAS Rietveld Refinement results for Mg(H₂O)₃B₁₂H₁₂.

Empirical formula	B ₁₂ H ₁₈ O ₃ Mg
<i>F</i> _w	220.18
Space group	R-3c (No. 167)
<i>a</i> (Å)	10.4597(12)
<i>c</i> (Å)	18.6077(21)
<i>V</i> (Å ³)	1763.1(5)
<i>Z</i>	6
<i>T</i> (K)	295
<i>λ</i> (Å)	1.540596
<i>ρ</i> _{cal} (g cm ⁻³)	1.21015(31)
pattern range (2 <i>θ</i> , °)	5 to 70
step size (2 <i>θ</i> , °)	0.0173
step scan time (s)	0.1
no. of contributing reflns	3757
<i>R</i> _{wp}	0.0709
<i>R</i> _p	0.0531
<i>R</i> (<i>F</i> ²)	0.1493

Table S4 Atomic coordinates for Mg(H₂O)₆B₁₂H₁₂.

Atoms	<i>x</i>	<i>y</i>	<i>z</i>	<i>occ</i>	<i>Beg</i>
B ₁₂ H ₂₄ O ₆ Mg					
Mg	0.5	0	0.5	1	3.8223
O(1)	0.5060(23)	0.9207(18)	0.2883(17)	1	2.6808
O(2)	0.7210(28)	0.1069(19)	0.4600(21)	1	6.3083
O(3)	0.3630(24)	0.1619(17)	0.4150(19)	1	6.6768
B(1)	0.0245(57)	0.0490(28)	0.1798(34)	1	-1.0378
B(2)	0.9834(63)	0.8382(45)	0.9709(43)	1	6.0366
B(3)	0.8922(48)	0.9141(36)	0.1308(40)	1	-0.7805
B(4)	0.8792(53)	0.0965(35)	0.8867(41)	1	1.0650
B(5)	0.8277(52)	0.0737(43)	0.0786(44)	1	2.6761
B(6)	0.8024(62)	0.9434(49)	0.9496(55)	1	4.9832
H(1)	0.04015	0.08031	0.294513	1	3.1214
H(2)	0.9728566	0.7350124	0.9524568	1	3.1214
H(3)	0.8233039	0.8592372	0.2141141	1	3.1214
H(4)	0.8022605	0.1580515	0.8145162	1	3.1214
H(5)	0.7178697	0.1206897	0.1288621	1	3.1214
H(6)	0.6762807	0.907284	0.9174626	1	3.1214

Table S5 Atomic coordinates for $\text{Mg}(\text{H}_2\text{O})_3\text{B}_{12}\text{H}_{12}$.

Atoms	<i>x</i>	<i>y</i>	<i>z</i>	<i>occ</i>	<i>Beg</i>
$\text{B}_{12}\text{H}_{18}\text{O}_3\text{Mg}$					
Mg	0	0	0.1940(1)	0.5	4.407
O	0	0.1688(14)	0.75	1	4.515
B(1)	0.1649(22)	0.1617(15)	0.016(1)	1	2.143
B(2)	0.9033(14)	0.9067(18)	0.071(1)	1	2.143
H(1)	0.291(13)	0.266 (11)	0.0277 (44)	1	2.143
H(2)	0.837(13)	0.846(11)	0.0966(52)	1	2.143

References:

S1. T. Peymann, C. B. Knobler, M. F. Hawthorne, *Inorg. Chem.*, 2000, **39**, 1163-1170.

S2. I. Tiritiris and T. Schleid, *Z. Anorg. Allg. Chem.*, 2004, **630**, 541-546.